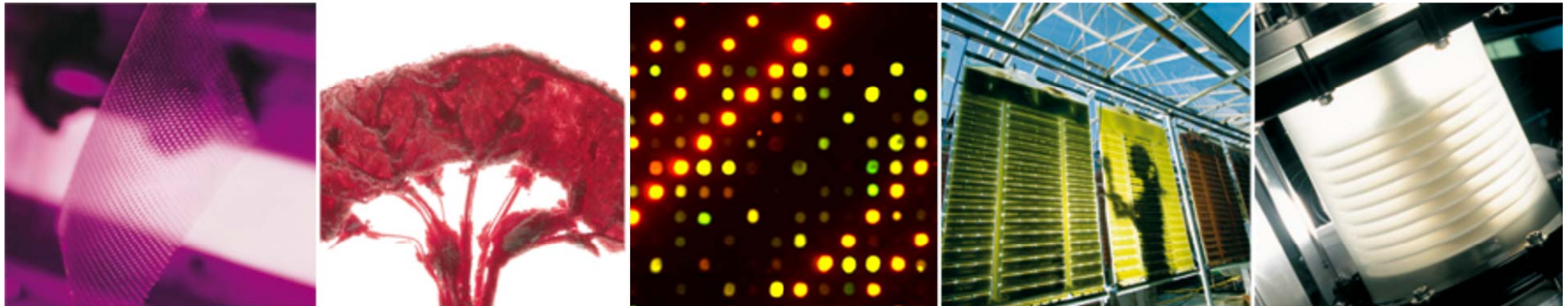

Contaminated Sediments – an Ecological Bomb

Removal and Reuse of Metal Sulfides from Water Using a Fixed-bed Anaerobic Loop Bioreactor

Dr. Werner Sternad and Dr. Dieter Bryniok

Water Workshop WATER QUALITY -
Novi Sad, 3rd September 2008



Bryniok, 09/08

Sediment Treatment

- dredging
- separation
 - gravel
 - sand
 - silt
 - clay
- de-watering
- **storage**
- further treatment
- utilization
- **final disposal**



Heavy Metals in Veliki Backi Canal Sediments at Vrbas

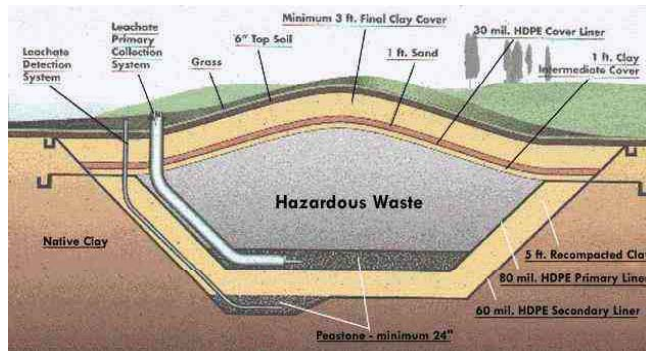


Most important contaminants:
heavy metals, **dissolved in water as metal ions** and/or **precipitated in the sediment as metal sulfides** and other hardly soluble salts

metal	average concentration (mg/kg)	maximum concentration (mg/kg)
Zn	1700	8500
Cu	280	620
Ni	74	170
Cr	200	1200

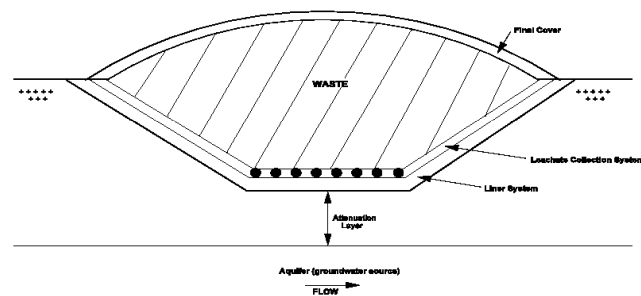
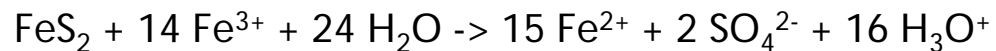
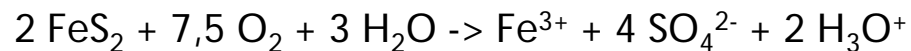
analyzed by University of Novi Sad, Faculty of Sciences, Department of Chemistry

Storage and Final Disposal of Sediments



oxidation of metal sulfides and mobilization of metals

- leachates formed by
 - humidity of the sediment
 - rainfalls
- abiotic oxidation (e. g. by O_2 or NO_3^-), favored by
 - low pH value
 - aeration and dissolved oxygen
 - carbonic acid
 - organic matter
 - light



- bacterial oxidation (e. g. B_d je jui j p c b d j m v t!
g f s s p p y j e b o t)
 - formation of sulfate favors additional abiotic oxidation

Avoiding the Release of Heavy Metals to the Environment

Sludge dredging and treatment:

- avoid oxygen transfer
- treatment of water before release back to the canal

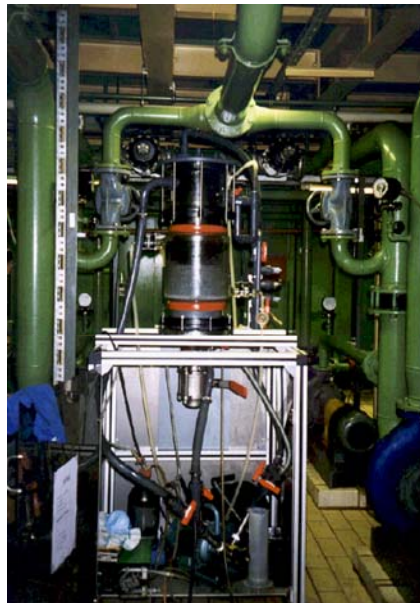
Storage of the sediments:

- disposal site with liner system and leachate collection and treatment
- prevent oxidation and leaching
 - covering the disposal site
- alternatively: treatment by bio-leaching of metals and treatment of the leachate
 - rinse the disposal site
 - aeration and pH adjustment
 - supply with organic substrates and HCO_3^-
- utilization of the sediment after treatment e. g. as building material

Final disposal at a safe landfill for hazardous wastes:

- disposal site with water-tight cover, liner system, leachate collection and treatment

Precipitation of Metals in a Fixed-bed Loop Bioreactor



Project at the Fraunhofer IGB

Treatment of wastewater from the automotive industry containing **cooling lubricants** and **heavy metals** in high concentrations

The concentration of the dissolved metals in the water phase had to be reduced below the legal limits

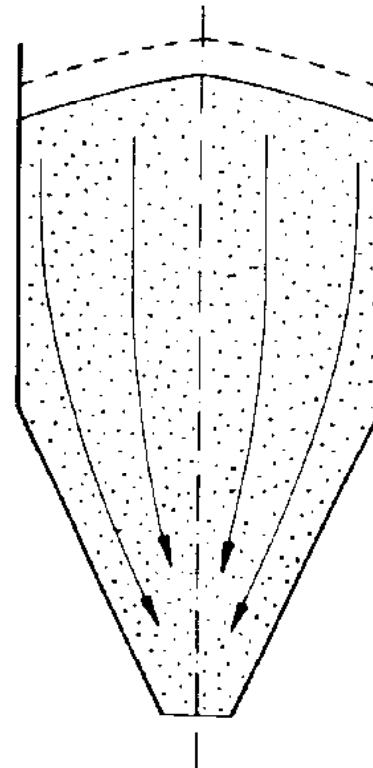
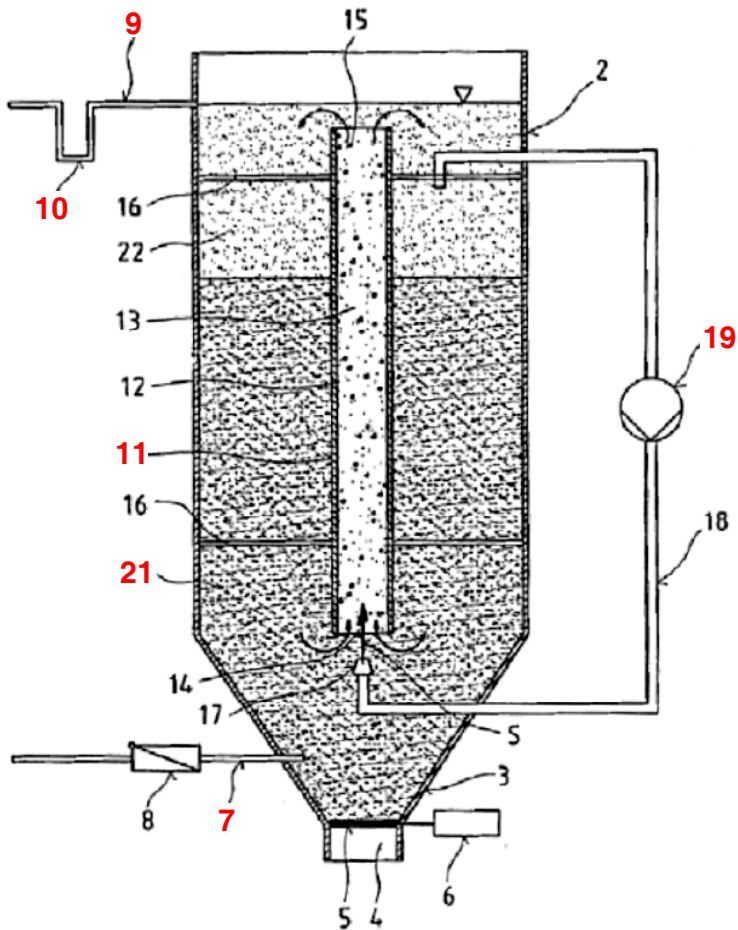
Solution:

Bio-precipitation of metal sulfides with sulfate-reducing microorganisms in a fixed-bed loop reactor

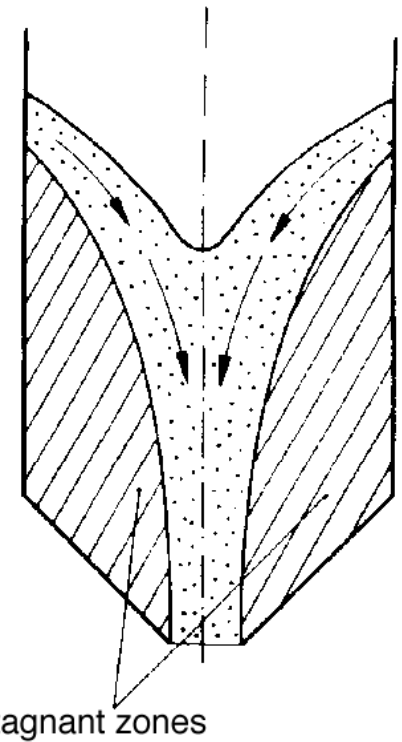
Removal of precipitated metal sulfides from carrier particles



Design of a Fixed-bed Loop Reactor (FBLR)



Mass flow



Funnel flow

Fixed-bed Loop Reactors



30 Liters



100 Liters

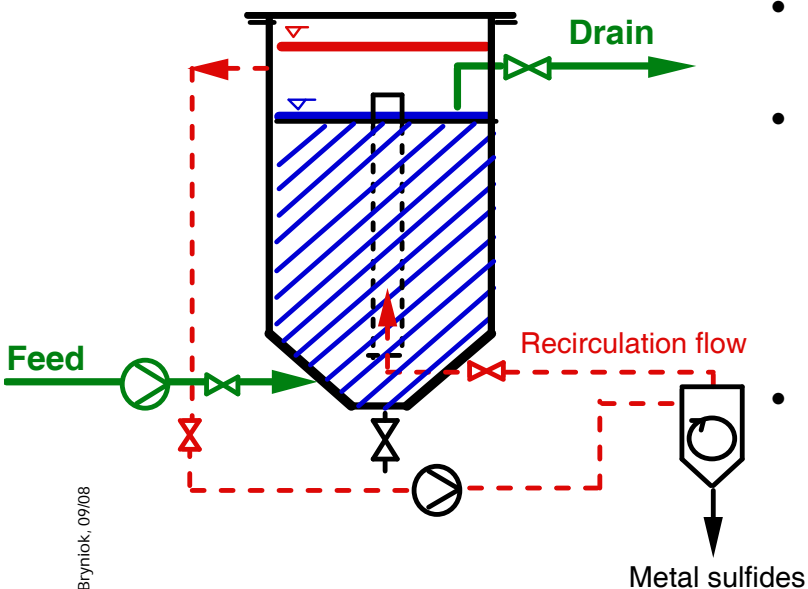


3500 Liters

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Principle of the Fixed-bed Loop Reactor (FBLR) Process

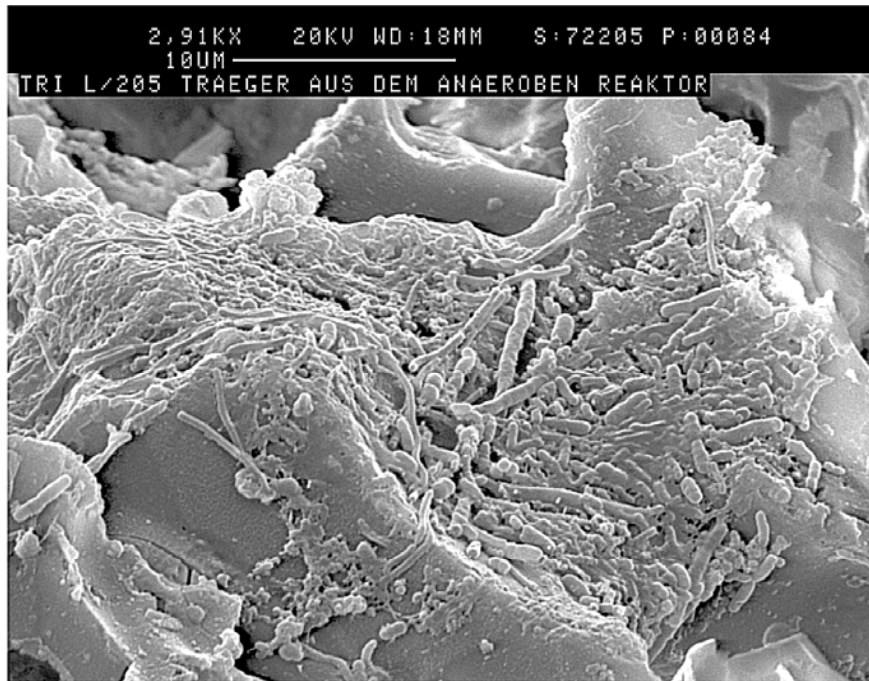
- Wastewater is fed into the bottom of the reactor and passes through the fixed bed.
- Sulfate-reducing microorganisms sulfate to sulfide and dissolved metals to insoluble metal sulfides.
- The fixed bed is loaded with the crystallized metal sulfides.
- Conventional fixed-bed reactors would block in course of time.
- A pump periodically drives water from the top of the FBLR into the inner conveying pipe. The particles of the fixed bed are transported to the top of the reactor. The resulting turbulence in the conveying pipe cleans the particles.
- Metal sulfides are separated in the hydro-cyclone.



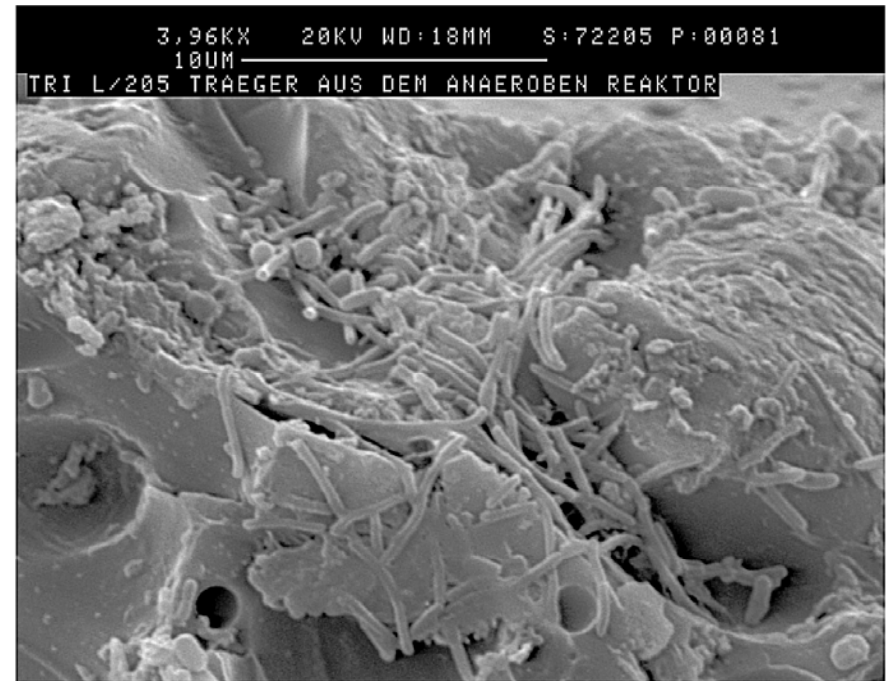
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Seite 9

Scanning electron micrograph of an overgrown carrier



3000 fold



4000 fold

Recovery of Metal Sulfides

hydro-cyclone



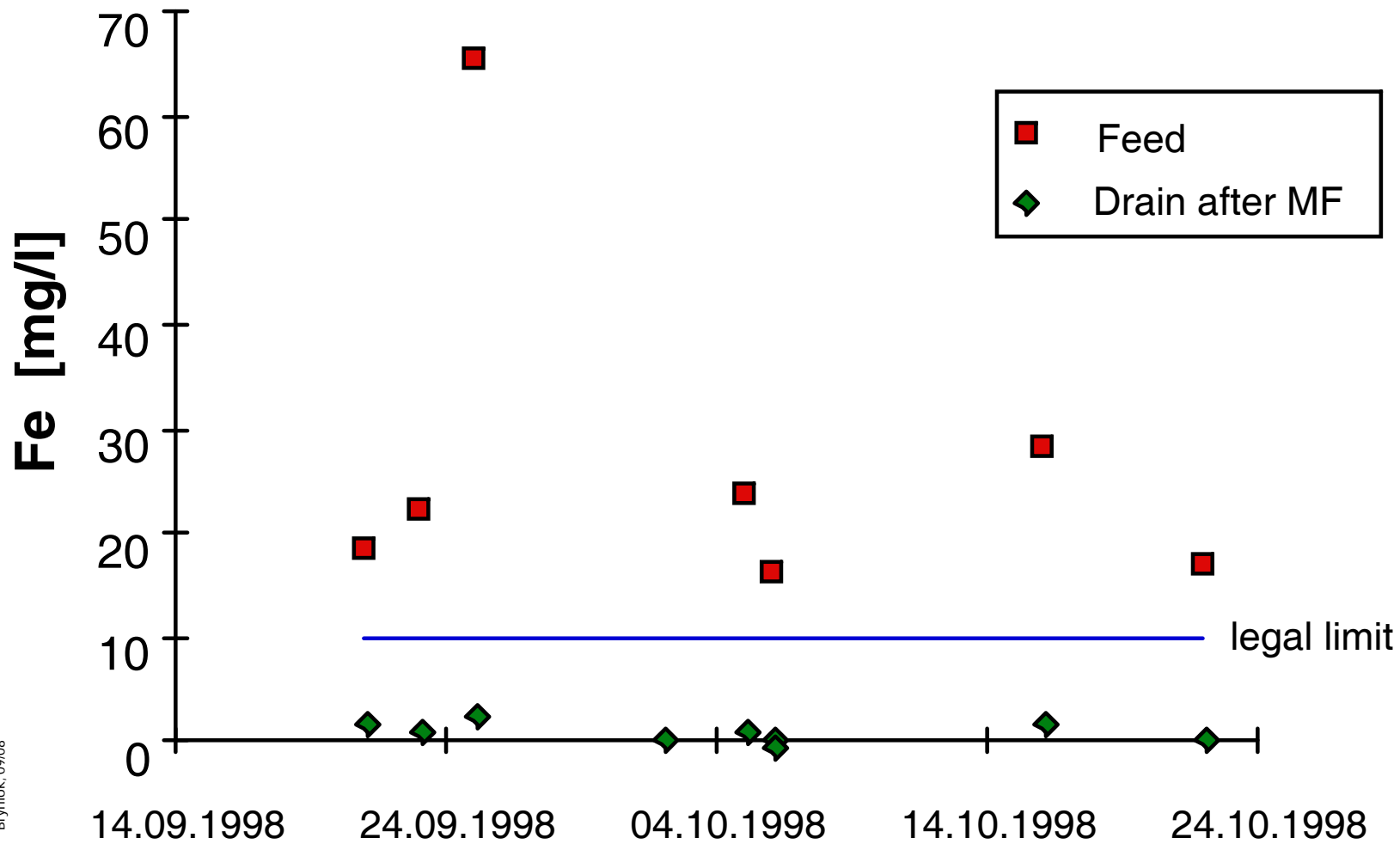
recovered sludge

biomass

metal sulfides

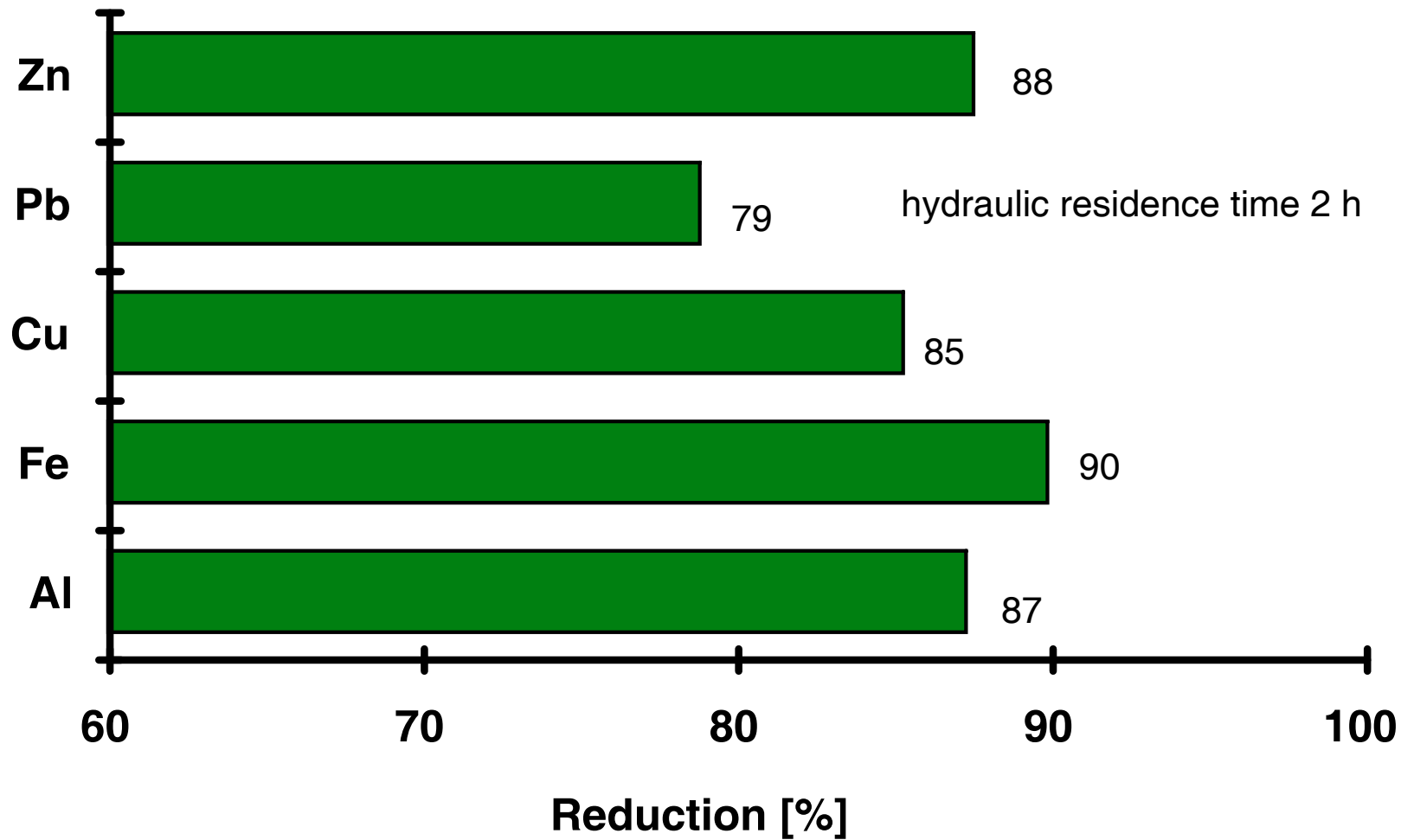


Concentration of Iron in Feed and Drain

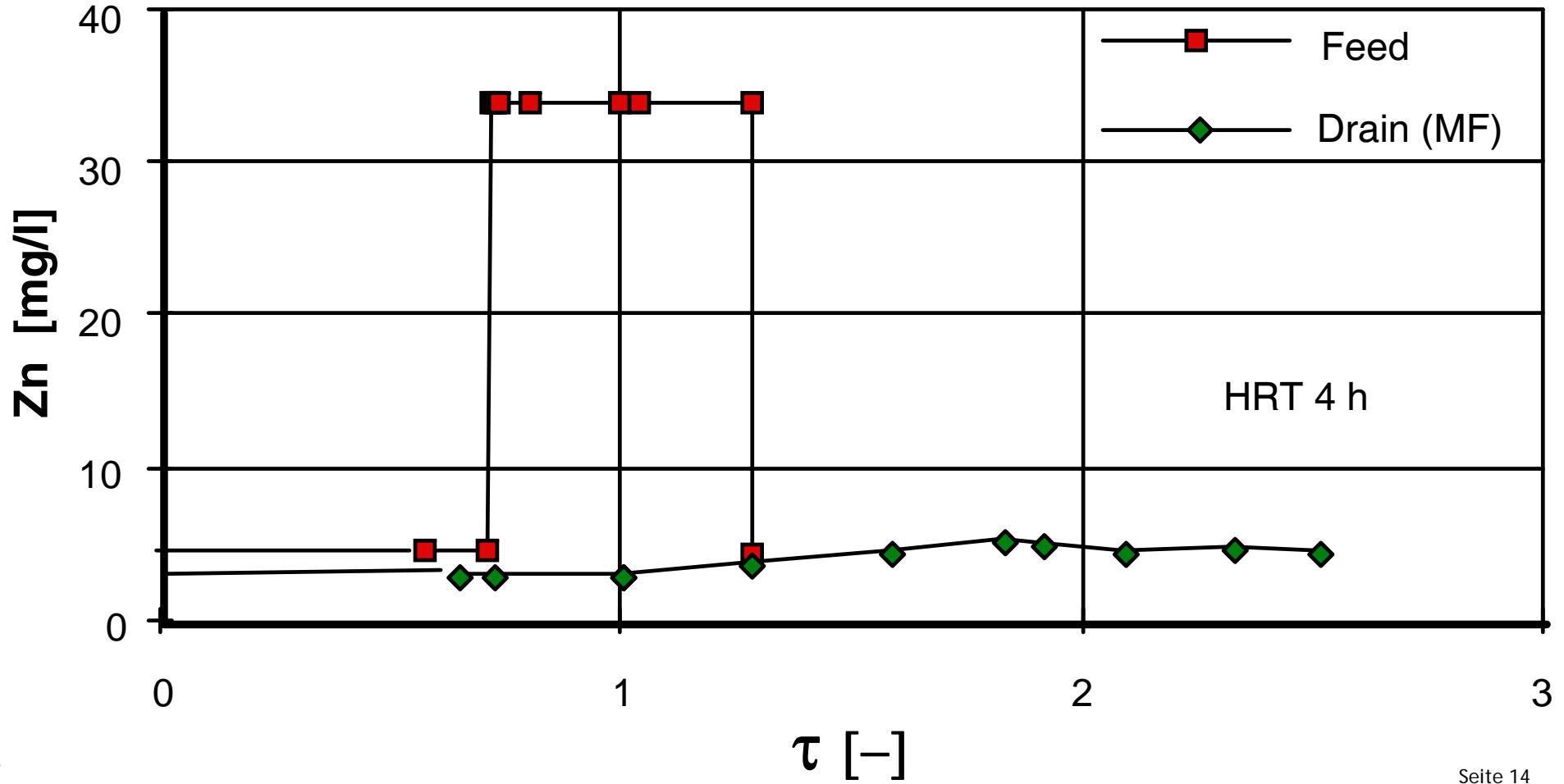


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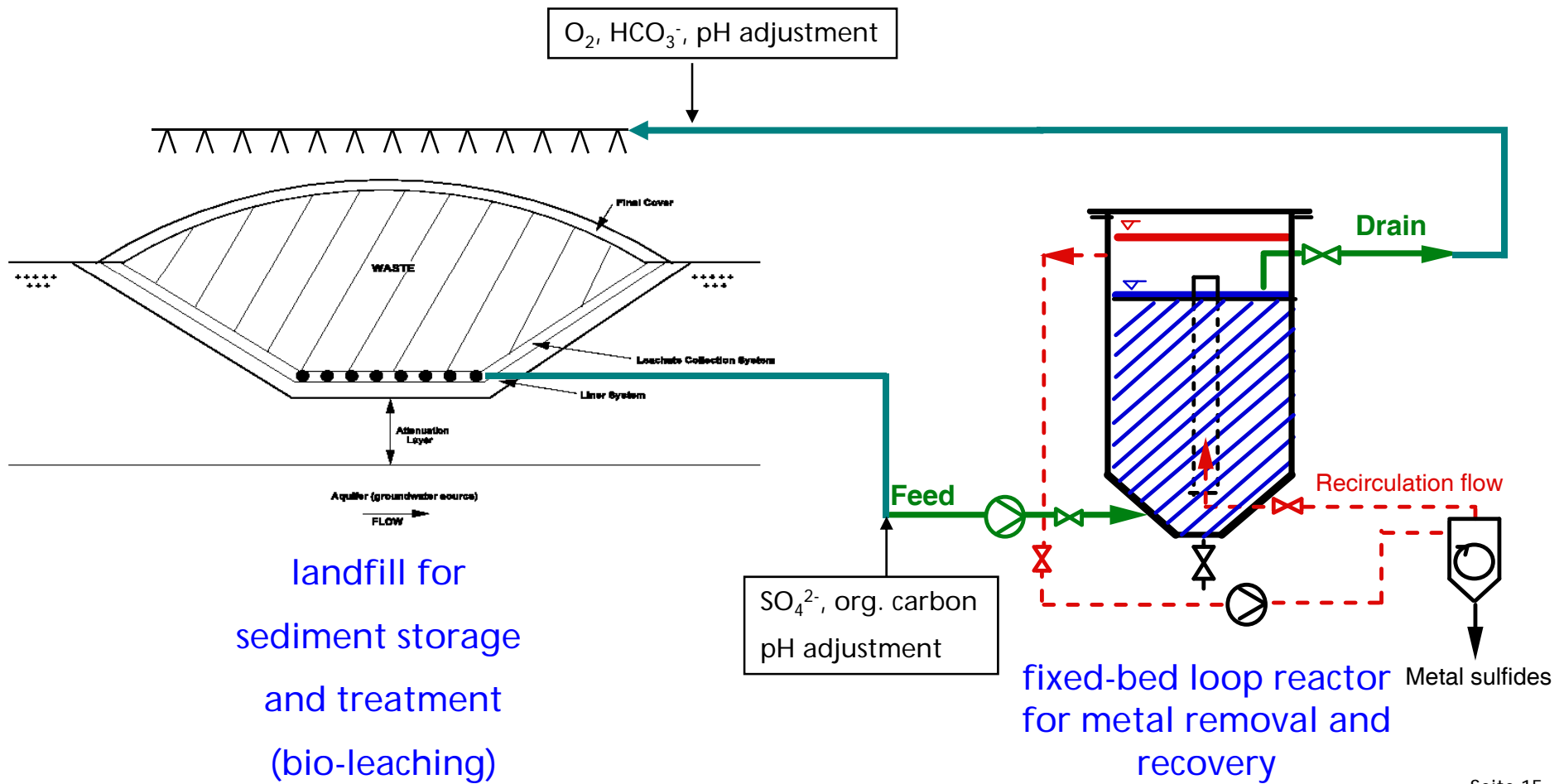
Reduction of Metal Sulfides in the Wastewater



Impact Load by Zinc



Perspective: Bio-leaching and Recovery of Heavy Metals



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Thank you very much for your attention!



any questions?

